Code: ME3T4

II B.Tech - I Semester–Regular/Supplementary Examinations November 2016

METALLURGY AND MATERIAL SCIENCE (MECHANICAL ENGINEERING)

Duration: 3 hours Max. Marks: 70

PART - A

Answer *all* the questions. All questions carry equal marks 11x = 22 M

1.

- a) Define Hardenability.
- b) Calculate packing factor of Face Centered Cubic.
- c) Write any two types of Bravais lattices with diagrams.
- d) Write Gibbs Phase rule and explain the terms in it.
- e) Explain peritectic reaction in Fe-Fe₃C equilibrium diagram.
- f) Write any four applications of low carbon steels.
- g) Draw the micro structure of Grey cast iron.
- h) What do you mean by sintering in powder metallurgy?
- i) Define Heat treatment and name various types of surface hardening processes.
- j) Explain Grain refinement.
- k) Write applications of composite materials.

PART - B

Answer any *THREE* questions. All questions carry equal marks. $3 \times 16 = 48 \text{ M}$

2.

- a) With neat sketches explain crystal imperfections. 8 M
- b) Define grain growth and enumerate the factors on which it depends. 8 M

3.

- a) Differentiate between substitutional and interstitial solid solutions with neat sketches. 6 M
- b) With a neat sketch explain Fe-Fe₃C equilibrium diagram. Show invariant reactions on it. 10 M

4.

- a) Explain types of cast irons with their micro structuresand applications.8 M
- b) Classify steels based on carbon content and also write the properties and applications of them with micro structures. 8 M

5.

- a) What is surface hardening? Name surface hardening processes. Explain any two of them in detail. 10 M
- b) Write short notes on Strain hardening and Solid solution strengthening. 6 M

6.

- a) Define Powder Metallurgy. Give advantages, limitations& applications of powder metallurgy.6 M
- b) Explain any three methods of manufacturing composites. 10 M